

**B.E. MECHANICAL ENGINEERING SECOND YEAR FIRST SEMESTER SUPPLEMENTARY
EXAM 2024**

MATERIAL SCIENCE AND ENGINEERING

Time: Three hours

Full Marks: 100

(Answer any FIVE questions)

- 1(a) What are the different crystal systems and crystal structures? Explain any two types of crystal system with suitable geometrical diagram. (6+4)
- 1(b) Explain how Miller indices are determined for planes of simple cubic crystal structure. (5)
- 1(c) Calculate the atomic packing factor for BCC crystal structure. (5)

- 2(a) What is diffusion co-efficient for solid state diffusion? Mention the factors which influence diffusion co-efficient. Explain how it is determined experimentally. (2+3+3)
- 2(b) What is decarburization? Draw the concentration profile for decarburization process and solve the following problem.

A 0.85% carbon steel component has been decarburised at 950°C for a duration of 5 hours in an atmosphere equivalent to 0.2% carbon at the surface of the component. Calculate the percentage concentration of carbon at a depth of 0.35 mm from the surface of the component. (4+8)

Given: $D_o = 0.7 \times 10^{-4} \frac{m^2}{s}$; $Q = 157 \frac{KJ}{mol}$; $R = 8.314 \frac{J}{mol K}$

Z	0.25	0.30	0.35	0.40
erf(Z)	0.2763	0.3268	0.3794	0.4284

- 3(a) What do you mean by slip system? Determine the number of slip system of FCC unit cell for {111} family of planes. (2+4)
- 3(c) Determine the magnitude of applied tensile stress which needs to be applied along $[1\bar{1}1]$ axis of a single crystal to cause slip on the $(1\bar{1}0)$ [101] system. $\tau_{CRSS} = 15 MPa$ (6)
- 3(c) Explain the concept of critical resolved shear stress. Show that yield stress is twice that of critical resolved shear stress. (4+4)
- 4(a) What is a composite material? Differentiate between composite material and alloy. Explain particle reinforced composite material with suitable examples. (2+4+4)
- 4(b) Derive the expression for 'critical length' of fibre in a fibre reinforced composite. Explain the stress-strain behaviour of FRC under longitudinal loading. (4+6)
- 5(a) Explain the origin of energy band structure in solids. (5)
- 5(b) Explain the Fermi-Dirac electron energy distribution function. Draw the function for temperatures 0 K, 300 K and 600 K for 'Silicon'. Also explain the dependence of conductivity on temperature with the help of the function. (3+3+3)
- 5(c) The resistivity of intrinsic semiconductor germanium at room temperature is 0.43 Ω -m. Calculate the intrinsic carrier density at room temperature and energy gap for germanium. (6)

Given data:

Number of electrons available for excitation near the top of valence band = $5 \times 10^{25} \text{ per } m^3$ Charge of electron = $1.602 \times 10^{-19} \text{ Coulomb}$; Mobility of electron = $0.39 \frac{m^2}{V.Sec}$ Mobility of hole = $0.19 \frac{m^2}{V.Sec}$; Boltzman constant = $8.62 \times 10^{-6} \frac{eV}{K}$

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- 6(a) Draw the Iron-Iron carbide equilibrium phase diagram according to scale and label it. Explain eutectic, eutectoid and peritectic reactions with reference to this diagram. (8+6)
- 6(b) 20 kg of an alloy with 60% lead and 40% tin is slowly cooled from 300°C. Refer to the lead-tin phase diagram given in Figure-1 and determine the followings: (6)
- (i) Weights of liquid phase and pro-eutectic solid phase just above the eutectic temperature
 - (ii) Weight of eutectic solid phase due to tin formed by eutectic reaction only
- 7(a) Draw the TTT diagram for eutectoid steel and mention the salient features of this diagram. (10)
- 7(b) Mention the objectives of heat treatment. How to perform normalizing and what are its effects on mechanical properties. (4+6)
- 8) Write short notes on the followings (any four): (4×5)
- a) HCP crystal structure
 - b) Piezoelectricity and its applications
 - c) Hume-Rothery's rule
 - d) Schmid factor
 - e) Fermi energy level
 - f) Lever rule
 - g) Burger vector

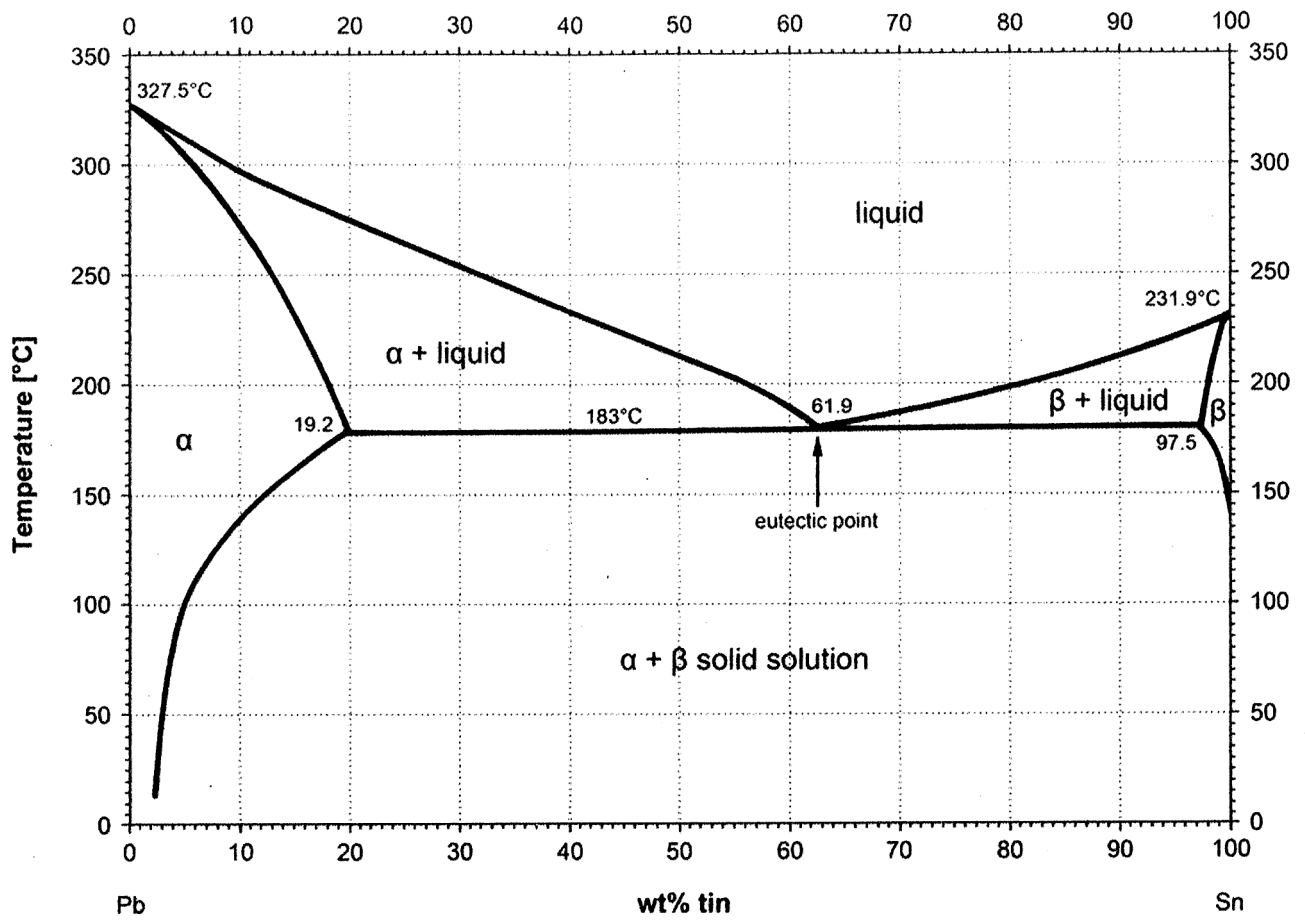


Figure-1: Pb-Sn phase