

Ref. No.: Ex/Che/PC/H/T/413/2024

BACHELOR OF CHEMICAL ENGINEERING EXAMINATION, 2024

(4th Year, 1st Semester)

INDUSTRIAL POLLUTION CONTROL ENGINEERING (HONS.)

Full Marks: 100

Time: 3 hours

Assume any missing data

Answer **20 marks** from Q. No. 1, **15 marks** from Q. No. 2, **25 marks** from Q. No. 3, **10 marks** from Q. No. 4 and all from Q.No. 5

Q.No.	CO	Question	Marks										
1.	2	<p>a. Describe the lime-soda ash process as water softening technique.</p> <p>A raw water to be treated by the lime soda process to the minimum hardness possible has the following characteristics:</p> <p>CO₂: 22 mg/L, Ca²⁺: 80mg/L, Mg²⁺: 12mg/L, Na⁺: 46 mg/L, HCO₃⁻: 152.5 mg/L and SO₄²⁻: 216 mg/L.</p> <p>For a flow of 25000 m³/day, calculate the chemical requirements and mass of solids and volume of sludge produced.</p> <p>Assume, the used lime is 90% pure and soda ash used is 85% pure.</p> <p>Also, assume that the specific gravity of the sludge is 1.04.</p>	5+15										
		<p>b. Find out a mathematical expression for active zone height in a fixed bed adsorber during breakthrough experiment.</p> <p>A breakthrough experiment is conducted for phenol producing results shown below: Determine the length δ of active zone. The diameter of the column used is 1inch and packed density of the bed is 721.58 kg/m³. [C₀] is equal to 25 mg/L. used</p> $\left(\frac{X}{M}\right)_{ultimate} = 0.02 \frac{kg\ phenol}{kg\ carbon}.$ <table><tr><th>C₀ (mg/L)</th><th>V (L)</th></tr><tr><td>0.06</td><td>1</td></tr><tr><td>1</td><td>1.24</td></tr><tr><td>6</td><td>1.31</td></tr><tr><td>10</td><td>1.43</td></tr><tr><td>15</td><td>1.48</td></tr></table>	C ₀ (mg/L)	V (L)	0.06	1	1	1.24	6	1.31	10	1.43	15
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		<table><tr><td>18</td><td>1.58</td></tr><tr><td>20</td><td>1.72</td></tr><tr><td>23</td><td>1.83</td></tr><tr><td>25</td><td>2.00</td></tr></table>	18	1.58	20	1.72	23	1.83	25	2.00									
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2.	1	<p>a. What is photochemical smog? How it causes formation of Peroxyacetyl nitrate (PAN)?</p> <p>b. What are the sources of lead in air as pollutant and what are the effects of it if present as air pollutant?</p> <p>c. What are different types of solid wastes?</p> <p>d. What are the effects of SO_x on environment as Criteria Air Pollutant?</p> <p>e. What are the reasons of drop of dissolved oxygen (DO) in water?</p> <p>f. Discuss the types of NO_x which are available as air pollutant.</p> <p>g. Why and how primary treatment of wastewater is done?</p>	<p>3</p> <p>3</p> <p>3</p> <p>3</p> <p>3</p> <p>3</p> <p>3</p>																
3.	4	<p>a. The results shown in the table below are to be used to design a standard conventional cyclone. The air stream is 21 m³/s and the diameter of the cyclone is 2.0 m. The temperature is 77°C and the specific gravity of the particle is 2.0. Assume that β equals 0.90. What are the diameter and terminal settling velocity of those particles that are 100% removed? Determine the expected percent removal of the particles.</p> <table><tr><td>Particle Size (μm)</td><td>Wt%</td></tr><tr><td>0 - 10</td><td>8</td></tr><tr><td>10 - 20</td><td>10</td></tr><tr><td>20 - 30</td><td>12</td></tr><tr><td>30 - 40</td><td>15</td></tr><tr><td>40 - 50</td><td>19</td></tr><tr><td>50 - 60</td><td>14</td></tr><tr><td>60 - 70</td><td>13</td></tr></table>	Particle Size (μm)	Wt%	0 - 10	8	10 - 20	10	20 - 30	12	30 - 40	15	40 - 50	19	50 - 60	14	60 - 70	13	25
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		<p>b. Measurement of the dust distribution of certain industrial operation yields the results in the table below. These results are to be used to design a plate type Electrostatic Precipitator (ESP). The air stream is 330 m³/s and the plates are spaced 30 cm apart. The voltage drop between discharge and collector plate is 70,000 V. The aspect ratio (length/width) is 1.2. The plate width is 10m and there are 80 channels. The temperature is 650°C and specific gravity of the particle is 2.0.</p> <p>i) what are the diameter and terminal settling velocity of the particles?</p> <p>ii) determine the expected percent removal of particles.</p> <p>iii) what is the power requirement assuming that the corona current is 2.3 A? Assume, β is 0.9</p> <table><tr><th>Particle Size (μm)</th><th>Wt%</th></tr><tr><td>0 - 10</td><td>8</td></tr><tr><td>10 - 20</td><td>10</td></tr><tr><td>20 - 30</td><td>12</td></tr><tr><td>30 - 40</td><td>15</td></tr><tr><td>40 - 50</td><td>19</td></tr><tr><td>50 - 60</td><td>14</td></tr><tr><td>60 - 70</td><td>13</td></tr><tr><td>70 - 80</td><td>9</td></tr></table>		Particle Size (μm)	Wt%	0 - 10	8	10 - 20	10	20 - 30	12	30 - 40	15	40 - 50	19	50 - 60	14	60 - 70	13	70 - 80	9	25
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4.	3	<p>a. Discuss the sequential steps of hard groundwater treatment to produce drinking water.</p> <p>b. What are the biological routes for solid waste treatment? Briefly discuss any one of them.</p>		10 10																		
5	1	<p>a. A perfectly mixed aeration pond with no recycle (return line)</p>		15																		

	<p>serves as the biological reactor for a small community. The pond receives 30 m³/d of influent with a BOD₅ of 350 mg/L that must be reduced 20 mg/L before discharge. It has been found that the kinetic constants for the system are $K_s = 100 \text{ mg/L BOD}_5$, $k_d = 0.1 \text{ d}^{-1}$, $\mu_m = 1.6 \text{ d}^{-1}$, and Y is 0.6 mg VSS/mg BOD₅.</p> <p>(i) What must the hydraulic detention time be in aeration pond?</p> <p>(ii) What mass of microbes will be produced in the pond each day?</p> <p>b. Derive the following for a centrifuge:</p> $t = \frac{18\mu}{D_p^2 \omega^2 (\rho_s - \rho)} \cdot \frac{h}{R}$ <p>Where, t = minimum retention time of solid having size $>D_p$ ω: angular velocity h: thickness of liquid layer in centrifuge bowl R: radius of centrifuge bowl</p> <p>Explain the scale-up criteria of a centrifuge.</p>	<p>15</p>
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