

**M. Sc. CHEMISTRY EXAMINATION, 2018**

( 4th Semester )

**PHYSICAL CHEMISTRY SPECIAL**

**PAPER - XV-P**

Time : Two hours

Full Marks : 50

( 25 marks for each unit )

Use a separate answerscript for each unit.

**UNIT – P- 4151**

Answer Question No. 1 and *any two* from the rest

1. Explain briefly the importance of “ *radial distribution function* ” to understand the structure of liquids. Find out the relation between “ *perturbed free energy* ” and “ *radial distribution function* ”. (Consider only first order density function).  $1\frac{1}{2} + 3\frac{1}{2}$
2. Considering Einstein’s model for the evolution of distribution of Brownian particles in space (1-dimension) and time, show that the root mean square displacement of Brownian particles is proportional to the *square root of time*. Explain the “ *dynamic equilibrium condition* ” for suspended Brownian particles irregularly dispersed in a liquid. For the Brownian system under the condition of dynamic equilibrium, obtain a suitable expression for the experimental verification of *Avogadro Number*.

5+2+3

[ Turn over

[ 2 ]

3. Deduce the Kramers equation for the Brownian motion with describing the significance of various terms appeared in the equation. Show that the equation obeys the Maxwell-Boltzmann distribution under equilibrium condition. Describe the essential postulates of Kramers rate theory for unidirectional chemical process. 4+3+3

4. Evaluate the theoretical basis for the existence of random force to describe the perpetual irregular motion of a Brownian particle of mass  $(m) = 2.5 \times 10^{-18} \text{kg}$  and diameter  $(r) = 2.5 \times 10^{-8} \text{m}$  in the medium of viscosity  $(\eta) = 0.01$  poise. What is meant by *drift* and *diffusion* terms for the Brownian motion in velocity space? Find out the expression for *diffusion coefficient* in velocity space for the Brownian particle of certain mass  $(m)$  in a medium with damping constant  $\alpha$  at temperature  $T$ .

4+2+4

### UNIT – P- 4152

5. Answer **any five** questions 5×5=25

- a) How does the hydrophobic interaction play an important role in protein structure? Discuss the role of solvent on the structure of a protein.
- b) How does a nerve impulse generate and what is the mechanism of its propagation?

[ 3 ]

- c) Why is sequence of DNA so important in a biological system? What is melting of DNA? How does melting temperature of DNA vary in presence of salt?
- d) Discuss the cooperative and non-cooperative binding of ligands to multiple sites of a bio-polymer. How can you explain electrostatic interaction in multiple equilibrium?
- e) What is stacking interaction? On which factors does stacking depend? Why is heparin more effective than DNA to stack dye molecules?
- f) What is the mechanism of muscle contraction?
- g) Write notes on -
  - i) Dynamic fluid structure of cell membrane.
  - ii) Synaptic transmission.