FINAL B. Sc. Examination, 2018

(2nd Semester, Special Supplementary)

CHEMISTRY (HONOURS)

PAPER - XVI

ORGANIC CHEMISTRY

Time: Two hours Full Marks: 50

Use a separate answerscript for each group.

GROUP-A

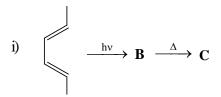
 a) Predict and classify (according to Baldwin) the two probable modes of cyclisation in the following amine A.
 Identify the disfavoured process with proper justification.

$$H_2N$$
 OEt OEt

b) (2*E*, 4*Z*, 6*R*)-2-Deutero-6-methyl-2, 4-octadiene on heating produces a mixture of two isomeric optically active acyclic conjugated dienes. Explain the observation.

 $2\frac{1}{2}$

c) Identify the products B, C, D and E of the following reactions with proper mechanistic and stereochemical interpretations on the basis of FMO theory. $2\frac{1}{2} + 2\frac{1}{2}$



ii)
$$\stackrel{\text{Ph}}{\longrightarrow}$$
 $\stackrel{\text{COOH}}{\longrightarrow}$ $\stackrel{\text{r.t.}}{\longrightarrow}$ $\mathbf{D} \text{ (Major)} + \mathbf{E} \text{ (Minor)}$
 $\downarrow 100^{\circ}\text{ C}$
 $\downarrow \text{Ph}$
 $\mathbf{D} \text{ (Minor)} + \mathbf{E} \text{ (Major)}$

d) Carry out the retrosynthetic analyses and hence suggest syntheses of the following compounds from readily available starting materials. $2\frac{1}{2}+2\frac{1}{2}+1+1$

ii) (utilizing three simple starting materials)

(from a ketone containing not more than five carbon atoms)

e)
$$Me$$
 $NaOMe, MeOH$

GROUP - C

5. Write the structure of expected major product in chair from for the following reactions. $1\frac{1}{2}\times 2$

i) D-Glucose $\xrightarrow{\text{Br}_2/\text{H}_2\text{O}}$

ii) D-Glucose $\xrightarrow{\text{MeOH/HCl}}$

6. How would you prove that structure of D-glucose contains six carbons in a straight chain? $2\frac{1}{2}$

7. Explain mechanistically why D-glucose and D-mannose give same product when treated with excess phenylhydrazine.

 $2\frac{1}{2}$

8. How would you convert D-aldopentose to D-aldohexose?

 $2\frac{1}{2}$

9. a) What do you mean by addition polymerization? Illustrate with suitable example.

b) Give one example of natural and synthetic rubber with their monomeric units. $1\frac{1}{2}$

10. What is the complementary nucleotide of adenine deoxyribonucleotide in DNA double helix? Give explanation in favour of your answer.

b) Depict a retrosynthetic pathway of the following compound:

4. a) Predict the product(s) and write the mechanistic pathway of the following reactions (any four): $1\frac{1}{2} \times 4$

b)
$$\xrightarrow{\text{Me}_2\text{CO},} \xrightarrow{\text{HCl}}$$

c)
$$\begin{array}{c} & & \xrightarrow{\text{PhCHO}} \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

$$\begin{array}{c} \text{H} \\ \text{d)} \\ \hline \text{mBuLi,PhH,rt.} \\ \hline \text{ii) nBuLi,PhH,rt.} \\ \hline \text{iii) H}_2\text{O, rt.} \\ \hline \text{iii) PhNO}_2/200^{^{\text{O}}}\text{C} \\ \end{array}$$

(mechanism is not necessary, only show the intermediates and find product)

iv) OH NH2

(from a starting material containing not more than six carbon atoms)

GROUP-B

- 2. a) Convert pyrrole to pyrrole-2-carboxaldehyde, and depict the mechanistic pathway for this transformation. 2
 - b) Outline the transformation of furan to 2-nitrofuran, and depict the corresponding mechanism.
 - c) Transform pyridine to 4-nitropyridine. $1\frac{1}{2}$
- 3. a) Carry out the synthesis of the following compounds from commercially available simple starting materials. $1\frac{1}{2}\times3$

iii) Quinoline (using glycerol as one of the starting materials).