B.E. CHEMICAL ENGINEERING FOURTH YEAR FIRST SEMESTER SUPPLEMENTARY EXAM 2018 (OLD) Bioprocess Engineering

Time: 3 hours Total Marks:100

Answer any four questions.

Use graph paper if required. Assume any missing data.

All symbols have usual significance

1.	Glucose is converted to ethanol by immobilized yeast cells entrapped in gel beads. The specific rate of ethanol production, q_p =0.2g ethanol/g cell-h. The effectiveness factor for an average bead is 0.8. Each bead contains 50g/L of cells. The voidage of the bed is 40%. Assume growth to be negligible (all glucose is converted to ethanol). The feed flow rate is 400L/h and the feed concentration of glucose is 150g/L. The diameter and the height f the column are 1m and 4m respectively and the yield coefficient is 0.49g ethanol/g glucose. What is the exit concentration of glucose? What is the concentration of ethanol in the exit stream? Show all relevant derivations.	25
2.	An industrial waste water stream is fed to a stirred tank reactor continuously and the cells are recycled back to the reactor from the bottom of the sedimentation tank placed after the reactor. The following information are given: F=100L/h; S0=5000mg/L; μ_m =0.25h ⁻¹ ; Ks=200mg/L; Recycle ratio=0.6; Degree of concentration in the sedimentation tank=2; $Y_{x/s}$ =0.4. The effluent substrate concentration is desired to be 100mg/L. Showing all derivations, a) Determine the reactor volume; (b) Determine the cell concentration in the reactor and in the recycle stream. (c) If the residence time in the sedimentation tank be 2h, determine the volume of the tank and cell concentration in the effluent of the sedimentation tank.	25
3.	Penicillin is produced by <i>P. chrysogenum</i> in a fed batch reactor with the intermittent addition of glucose solution to the culture medium. The initial culture volume at quasi-steady state is 500L and glucose containing nutrient solution is added with a flow rate F=50L/h. Glucose concentration in the feed solution and initial biomass concentration are 300g/L and 20g/L respectively. The kinetic constants and yield coefficient of the organism are μ_m =0.2h ⁻¹ , K_s =0.5g/L and $Y_{x/s}$ =0.3g /g glucose.Showing all derivations a) Determine the concentration and total amount of cells and products at t=10h if q_p =0.05 g product/g cell-h and C_{p0} =0.1g/L.	25
4.a)	Clarified bioreactor broth contains a protein at a concentration of 15g/L. Product is harvested from the broth using ultrafiltration at a fluid velocity of 0.34m/s in open membrane tubes of diameter 24mm and length of 2m. Deriving the necessary equation, estimate the permeate flux if the filter is operated under gel polarization condition and protein concentration in the gel is 25 g/L. The properties of bioreactor broth: ρ =1020kg/m³; μ =1.8cP; D=3.6x10 ⁻¹¹ m²/s .[N _{Sh} =0.023(N _{Re}) ^{0.89} (N _{Sc}) ^{0.3}]	15
4.b).	A protein solution contains 1.95M KCl. Constant volume diafiltration using a membrane with retention coecfficient R=0 for KCl and R=1 for protein is used to desalt 2000L of the protein solution. The filter is operated so that the permeate flux is 20L/m²-h. The total membrane	

Co e	equation and deter				Derive the governing olume of permeate		
id.) g	generated.						
	Carbohydrate A d	lecomposes in t	the presence of	enzyme E. We	also suspect		
1	that carbohydrate B in some way influences this decomposition. To study this phenomenon various concentrations of A, B, and E flow into and out						
				B, and E flow	into and out		
1	of a mixed flow reactor $(V = 240 \text{ cm}^3)$. (a) From the following data find a rate equation for the decomposition.						
	(a) From the foll-	owing data flu	d a rate comati	on for the deco	omposition.		
	(a) From the folk (b) What can you						
1		i say about the	role of B in t	he decompositi			
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1	(b) What can you (c) Can you sugg $\frac{C_{A0}, \text{ mol/m}^2}{200}$	is ay about the est a mechanis $C_{\rm A}$, ${ m mod/m^3}$	e role of B in to sm for this read C_{180} , $ m mol/m^3$	he decomposition? $C_{\rm E0}$, ${ m mol/m}^3$ 12.5	on ⁷ v, cm³/min 80		
	(b) What can you (c) Can you sugg $\frac{C_{A0}, \text{mol/m}^2}{200}$	est a mechanis $\frac{C_{\text{A}}, \text{ mol/m}^{3}}{50}$ 300	e role of B in the seminary for this read $C_{(80)}$, $\mathrm{mol/m^3}$ 0	the decomposition? $C_{\rm E0}, { m mol/m^3}$ 12.5 5	o, cm³/min 80 24		
1	(b) What can you (c) Can you sugg $C_{A0}, \text{mol/m}^2$ 200 900	a say about the est a mechanis $C_{\rm A}$, mol/m ³ 50 300 800	e role of B in term for this read $C_{(80)}$ mol/ m^3 0 0	the decomposition? $C_{\rm Eir}$, ${ m mol/m^3}$ 12.5 5 5	v, cm³/min 80 24 48		