M.E. CHEMICAL ENGINEERING FIRST YEAR FIRST SEMESTER – 2018 INTERFACIAL SCIENCE AND ENGINEERING

Time: 3 hrs Full Marks: 100

Answer any Four questions

- 1. (a) The critical coagulation concentrations for NaCl, MgCl₂ and AlCl₃ for negatively charged As₂S₃ colloids are 70 mol/m³, 0.8 mol/ m³ and 0.09 mol/ m³, respectively. Verify whether these values are consistent with the Schulze- Hardy rule or not. (5)
 - (b) Derive the expression for terminal velocity based on Stokes' law. (10)
 - (c) A spherical particle suspended in water is placed in a centrifugal field. The diameter of the particle is $1*10^{-7}$ m. What should be the rotational speed so that the particle moves from 6.5 cm to 7 cm in 60 s? Density of the particle is 7500 kg/m³. (10)
- 2. (a) Define Donnan Equilibrium. State the significance of Deborah number and Peclet Number. (5+5)
 - (b) The aggregation number of sodium dodecyl sulfate micelle in water is 100. Calculate the packing parameter, and predict the shape of the SDS micelles. (10)
 - (c) Define Kraft point and cloud point. (5)
- 3. (a) Estimate the surface tension of ethyl acetate at 298 K using the parachor date. Given the density of ethyl acetate is 0.902 g/cc. State the importance of Harkins-Brown factor (6+4).
 - (b) The interfacial tension between an oil and water is 50 mN/m. The density of oil is 850 kg/m³. If the angular velocity is 1050 rad/s, calculate the radius of the cylindrical drop.(7)
 - (c) Explain the working principle of tapping mode atomic forced microscopy. (8)
- 4. Write Short notes on (any five): (i) Scanning Electron Microscopy; (ii) Anionic Surfactants; (iii) Fowkes correlation; (iv) Advancing and Receding Contact Angle; (v) Lennard-Jones potential; (vi) Electrostatic double layer. (5*5)
- 5. (a) Explain the decaying profile of Potential as a function of distance considering Debye-Huckel Approximation. (10)
 - (b) Show that the ratio of force corresponding to VDW interaction energy between a sphere (radius = Rs) and planar half-space (F_{sp}) and Plane parallel halfspaces (F_{pp}) do not depend on Hamaker constant. (10)
 - (c) Differentiate hydration force from hydrophobic interaction. (5)