M. Sc. (BIOTECHNOLOGY) Examination, 2023

(1st Year, 2nd Semester)

BIO-ANALYTICAL TECHNIQUES

PAPER – 236

Time: Two hours Full Marks: 50 (Written 40 + Internal Assessment 10)

Group - A

Answer any two questions.

1. Answer *any five* questions :

 $5 \times 1 = 5$

- a) What is Radioactive Decay?
- b) Define Isotope? Give an example of isotope.
- c) Possibly how many kinds of different radiations are coming off from radioisotope?
- d) Among different types of radiations which one has the highest energy?
- e) Which radioisotpes are most frequently used in the biological experiments?
- f) Name the technique by which the 5'-end of a DNA molecule can be labelled.
- g) Which nucleotide is typically used to specifically label RNA?
- h) Which of these radioisotopes, either ¹⁴C or ³²P emits β particles of higher energy?

2. Write Short Notes on:

 $2 \times 2.5 = 5$

- a) Decay of β particle.
- b) Geiger-Muller Counter
- c) Different Methods of Labelling DNA
- d) Southern Blotting Technique
- 3. a) Name a technique which is used to detect radioactivity in a quantitative fashion.
 - b) What is the 'half-life' (in days) of the following isotopes: ³H, ³²P, ³⁵S and ¹⁴C.
 - c) If you use ^{32}P (emits β -particles) in a DNA-labelling experiment. What kind of protection you would exercise a lead shield or a perspex shield? 1+2+2=5
- 4. a) If you want to label a piece of DNA, which one of the following isotopes you will use ¹⁴C or ³²P? Justify your answer.
 - b) Are the energy level of β -particles emitted from various radioisotopes (such as 3 H, 14 C and 32 P etc.) same?
 - c) Mention three applications of radio isotopic methods in various fields of biology, medicine and agriculture. 1+1+3=5

[Turn over

5. Mention if following statements are true or false:

 $1 \times 5 = 5$

- i) Alpha particles are hydrogen nucleus.
- ii) Beta particles are equivalent to electrons.
- iii) Liquid Scintillation Counter is used to detect gamma rays.
- iv) ¹⁵N is radioactive isotope.
- v) Gamma rays have the highest penetrating capacity among all kinds of radiations.

Group - B

Answer any two questions.

- 6. What is the main use of confocal microscope? Why pinhole is used in confocal microscopy? Why are lasers used in confocal? 1+2+2
- 7. What is the principle of FACS? Outline briefly the method to get a cell cycle profile for normal cell? Draw a cell cycle graph for normal cells and G1 arrested cells? 1+2+2
- 8. Draw a schematic diagram of TEM indicating different components. Are same microscopy techniques are employed in SEM and TEM? Why Glutaraldehyde and osmium tetroxide are used in TEM?

 2+1+2
- 9. Mention the basic principle of Atomic forced microscopy (AFM)? What are the advantages of AFM? What is the principle of phase contrast microscope? 2+1+2

Group - C

10. Answer *any two* questions.

3 marks each

- a) Quartz cuvettes are much more expensive than glass cuvettes. Then why do spectrometer users use quartz cuvettes?
- b) What are the ranges of wavelength of UV and visible lights?
- c) Among UV and visible lights, which one has a higher level of energy?
- d) Why is an extremely low value of absorbance in the order of 0.001 not acceptable?
- e) A solution shows an O.D. of 0.455 at 280 nm in a certain spectrometer. If other spectrometers are used, do you expect the same O.D. of the same solution at the same wavelength? (Within limits of error)

11. Answer *any two* questions.

2 marks each

- a) In SDS PAGE, globular proteins are separated based on which property? In non-denaturing PAGE, proteins are separated based on which properties? [No explanation is needed.]
- b) Holding all experimental conditions constant, if the degree of cross linking is increased in SDS-PAGE, the rate of migration of proteins will be faster or slower?

- c) How will you calibrate a 1000 microlitter pipetman?
- d) Among a spectrometer and a spectrofluorimeter, which one is more sensitive?
- e) In a laboratory, where will you store inflammable organic solvents near a window or near the exit or away from the exit, etc? In a laboratory, how many exit doors are recommended?

Group - D

12. Answer *all* the questions.

a)	In which frequency range NMR and ESR are observed?	1
b)	Define the term Luminescence.	2
c)	Phosphorescence is the emission of light from singlet excited states. Is this statement true false. Justify your answer accordingly.	e or 1
d)	Define Circular Dichroism.	2
e)	What is the principal use of CD?	1
f)	What is meant by Larmor Procession?	1
g)	Explain the phenomenon of NMR in one sentence.	2