

2<sup>ND</sup> YEAR POWER ENGG. 2<sup>ND</sup> SEMESTER EXAMINATION, 2017

## SUBJECT: Engineering Thermodynamics-II

Time: Three Hours

Full Marks 100

Answer question No. 1 and any FIVE from the rest.

No. of Questions	Use of steam Table, Refrigerant R134a and enthalpy table allowed	Marks
1.	a) Why the compression ratio of a CI engine is more than that of SI engine?	2
	b) Explain the utility of a reheater in a reheat rankine cycle.	4
	c) Why expander is used in gas refrigeration instead of throttling device? Explain with temperature inversion curve.	4
	d) How will the Relative Humidity and Humidity Ratio of the air contained in a closed room changes as it is cooled?	2
	e) In a Brayton cycle with air ( $\gamma = 1.4$ ) as working fluid, $T_{\min} = 300$ K and $T_{\max} = 1000$ K. Determine The pressure ratio corresponding to maximum net work per cycle.	2
	f) The maximum pressure and temperature in an air standard diesel cycle are 44 bar and 1600 K, respectively. If the minimum pressure and temperature are 1 bar and 300 K, respectively, then determine the cut-off ratio.	4
	g) What is equivalence ratio related to combustion of gas mixture? Explain rich and lean mixture in terms of equivalence ratio.	2
2.	a) Compare air standard efficiency of Otto, Diesel and dual cycle for same maximum pressure and temperature and heat rejection with p-v and T-s diagrams.	5
	b) Show an air standard Otto cycle on p-v and T-s diagram clearly stating the processes. Derive its thermal efficiency in terms of compression ratio.	8
	c) An engine operating on Otto cycle having cylinder bore of 15 cm and stroke of 45 cm. If the clearance volume is 2000 cm <sup>3</sup> , compute the air standard efficiency.	3
3.	a) A gas-turbine power plant operating on a Brayton cycle has a pressure ratio of 6. The gas enters the compressor at 0.1 MPa, 27°C. Maximum cycle temperature is 927°C. Assuming a compressor and turbine efficiency of 90% each, determine (a) the back work ratio, (b) the gain in thermal efficiency due to regeneration, and (c) the turbine exit temperature.	16
4	A steam power plant operates on the reheat Rankine cycle. Steam enters the high-pressure turbine at 12 Mpa and 500°C at a rate of 8 kg/s and leaves at 2 MPa. Steam is then reheated at constant pressure to 450°C before it expands in the low-pressure turbine at 10 kPa. Steam leaves the condenser as a saturated liquid. Determine (a) the net power output, and (b) the thermal efficiency of the cycle.	16
5.	a) Cooling Water leaves the condenser of a power plant and enters a cooling tower at 35°C at a rate of 100 kg/sec. Water is cooled to 22°C by the air entering at 1 atm 20°C and 60% relative humidity. Air leaves the cooling tower at 30°C saturated. Determine the volume flow rate of air and amount of makeup water.	12
	b) What is adiabatic saturation process? Show the Process on T-s diagram?	4
6.	a) An ideal gas refrigeration cycle using air as the working fluid is to maintain a refrigerated space at $-23^{\circ}\text{C}$ while rejecting heat to the surrounding medium at 27°C. If the pressure ratio of the compressor is 3, determine (a) the maximum and minimum temperatures in the cycle, (b) the coefficient of performance, and (c) the rate of refrigeration for a mass flow rate of 0.08 kg/s.	8

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b)	A refrigerator uses refrigerant-134a as the working fluid and operates on an ideal vapor-compression refrigeration cycle between $-20^{\circ}\text{C}$ and $0.9\text{ MPa}$ . The mass flow rate of the refrigerant is $0.05\text{ kg/s}$ . Show the cycle on a $T$ - $s$ diagram with respect to saturation lines. Determine (a) the rate of heat removal from the refrigerated space and the power input to the compressor, (b) the rate of heat rejection to the environment, and (c) the coefficient of performance.	8
7. a)	Considering the 1 <sup>st</sup> $Tds$ equation and 2 <sup>nd</sup> $Tds$ equation show that $\therefore (c_p - c_v) = \frac{Tv\beta^2}{K_T}$ where $\beta$ = volume expansivity and $K_T$ = isothermal compressibility	10
b)	What is Joule Thomson Coefficient? Write the expression for Joule Thomson coefficient and determine its value for ideal gas.	6
8 a)	One kmol of ethane ( $\text{C}_2\text{H}_6$ ) is burned with an unknown amount of air during a combustion process. An analysis of the combustion products reveals that the combustion is complete, and there are 3 kmol of free $\text{O}_2$ in the products. Determine (a) the air-fuel ratio and (b) the percentage of theoretical air used during this process (c) Assuming complete combustion and a total pressure of 1 atm, determine the dew-point temperature of the products.	12
b)	Define enthalpy of formation.	2
c)	What is adiabatic flame temperature? When its value will be maximum?	1+1