B Pharmacy First Year 2nd Semester Examination 2017

Pharmaceutical Chemistry - I (Organic-1)

Answer Five questions taking at least two from each group

GROUP - A

Time - 3 Hours Full marks- 100

Q1. a) Deduce the structure of benzene.

- 10

- b) Show reaction mechanism for aromatic electrophilic substitution reactions. i) Nitration
 - ii) Friedel craft's Reaction

 $5 \times 2 = 10$

- Q2. i) Explain rules of orientation in benzene.
 - ii) Mention three groups each of ortho/para and meta orienting
 - iii) Give the structure of p-iodo-o-cresol & 3-bromo-4-hydroxybenzoic acid.
- iv) What will be the product when benzene is treated with chlorine in presence of iron and in presence of sunlight only.
 - v) When o-chloronitrobenzene is treated with alkali what will be the product.
 - vi) How do you prepare trifluoromethylbenzene from toluene
 - vii) What will be the product when p-hydroxytoluene is treated with bromine
 - ix) Mention the product of nitrobenzene in presence of triphenylphosphine.
 - x) Show the rearrangement of phenylhydroxylamine to p-aminophenol.

 $2 \times 10 = 20$

- Q3. a) i) How do you prove that during nitration alpha complex formation takes place before final product of nitrobenzene.
 - ii) How do you prepare o-dinitrobenzene
 - iii) Show preparation of benzylideneaniline from aniline.
 - iv) Write reaction of aniline with carbon disulphide and alkali.
 - v) Among ortho and meta nitroaniline which one gives nitrophenol with alkali.

 $2 \times 5 = 10$

b) Show the reaction steps for the synthesis of following compounds.

 $2 \times 5 = 10$

N-methylformanilide, Michler's ketone, Diphenylamine, Benzimidazole. Quinoxaline

B. Pharm. 1st Year 2nd Semester 2017

PHARM. CHEM. I (Organic-I)

Time: 3 h

Full Marks: 100

Group 'B'

Answer at least one question from this group.

- 4. a) Write a note on nomenclature of dihydric alcohols.
- b) How do you prepare glycerol? Discuss synthetic steps of preparations using chemical equations.
 - c) Write structures and names of oxidative products of ethylene glycol.

6 + 8 + 6 = 20

- §. a) Discuss the general methods of preparation of ketones mentioning the difference between these from the general methods of preparation of aldehydes.
- b) Write structures and names of oxidative products of glycerol.

12 + 8 = 20