B. MECH ENGG. FIRST YEAR, 2ND SEMESTER EXAMINATION 2017

THERMODYNAMICS

Time: Three hours Full Marks: 100 Answer question No. 1 (Compulsory) and any four questions from the rest

Answer to all parts of a question must be together

NB: Assume any data, if not furnished, consistent with the problem. Use of relevant tables and charts are permitted.

- 1. a) Compare heat and work highlighting their similarities and differences.
 - b) Show the following processes for a pure substance with proper labeling:
 - i)) Isothermal process from compressed liquid zone to superheated vapor zone on p-v plane.
 - ii) Isobaric process from superheated vapor zone to sub-cooled liquid zone on T-s plane.
 - c) Prove that $pv^k = constant$ for an adiabatic process where k is the ratio of the specific heats.
 - d) A cyclic engine operating between two thermal reservoirs has a thermal efficiency of 0.5. What is the COP of the same system if reversed to use as a refrigerator?
 - e) Define thermal equilibrium

4+6+5+3+2

- 2. a) Explain sublimation process with reference to the p-T and T-v diagram.
 - b) Explain 'critical point' with proper diagrams.
 - c) In the turbine of a gas turbine unit the gases flow through the turbine at 17kg/s and the power developed by the turbine is 14 MW. The specific enthalpies of the gases at inlet and outlet are 1200 kJ/ kg and 360 kJ/ kg respectively with corresponding velocities at 60m/s and 120 m/s. Calculate the rate at which heat transfer takes place from the turbine, Find out the diameter of the inlet pipe if the specific volume at inlet is $0.5 \text{ m}^3/\text{ kg}$.

4+4+12

- 3. a) State the first law of Thermodynamics for a system undergoing a non-cyclic process. Prove that internal energy is a property of the system.
 - b) Steam at 0.6 MPa, 200⁰ C enters an insulated nozzle with a velocity of 50 m/s. It leaves at a pressure of 0.15 MPa and a velocity of 600 m/s. Determine the final temperature if the steam is superheated in the final state or the quality if it is saturated.

8+12

- 4. a) Show that no heat engine can have higher efficiency than that of a reversible engine for given two heat reservoirs.
 - b) Two Carnot engines A and B are operating in series. a rejecting heat directly to B. Engine A receives 200 KJ at a temperature of 421° C. Engine B rejects heat at temperature 4.4° C. If the work output of A is twice that of B, find i) intermediate temperature between A and B, ii) efficiency of each engine, iii) heat rejected to the cold sink

~8+12

- 5. a) Establish the Maxwell Relationships.
 - b) Derive the Clapeyron equation
 - c) Explain the isenthalpic expansion of a real gas for different inlet pressure and temperature with the significance of inversion curve.

8+6+6

- 6. a) A 1 kg sample of moist air initially at 21[°]C and 1 bar with 70 % relative humidity is cooled to 5[°]C at constant pressure. Determine the initial humidity ratio, dew point temperature and the amount of water that is condensed during the process.
- b) Calculate the stoichiometric air-fuel ratio of the combustion of coal of following composition by mass; C 90%, H₂ 3%, O₂ 2.5%, N₂ 1%, S 0.5%, Ash 3%.

10 + 10

- 7. Write short notes on any four (4):
 - a) Entropy generation
 - b) Thermodynamic wet bulb temperature
 - c) Throttling process
 - d) Carnot cycle
 - e) p-v-T surface
 - f) Exergy of open system

5 x 4