B.Pharm. 3rd year 2nd semester examination, Session 2021-22

Sub: Pharmaceutical Engineering I Ref: Ex/Pharm/T/325

Total Marks 100 Time 3 hour

(Each question carries 20 marks. Answer any 5 questions)

Q1. (a)Derive mathematical expression for Steady State Heat Flow by conduction through a series of layers.

- (b) Derive overall heat transfer coefficient from individual coefficients when heat is transferred from one fluid to another fluid via tube wall.
- (c) Wall of a flat furnace is constructed of two layers of bricks. The temperature of the inner face of wall is 760°C and that of outer wall is 70°C. The first layer (thickness =11.4 cm, k=0.138 W/m-°C) is backed by the 2nd layer (thickness =23 cm, k=1.38 W/m-°C).(a) what is the heat loss through the wall?, (b) What is the temperature between two layers?

Marks: 6+8+6

- Q2. (a)Describe function of a long tube evaporator (climbing film) with a neat diagram.
- (b) A single effect evaporator is to concentrate 9070 kg/h of a 20% sodium hydroxide to 50% solids. The gauge pressure of the steam is to be 20 psi (214.96°C), the absolute pressure in the vapor space is to be 100 mm Hg, (enthalpy of superheated vapor at liquor side (temp.91.66°C), λ =2672.54 kJ/kg) and heat of vaporization of steam λ s at gauge pressure of 20 psi is 2184 kJ/kg. The boiling point of solution is 91.66°C. The overall coefficient is 1400 W/m²-°C. The feed temperature is 37.8 °C and its enthalpy is 127.93 kJ/kg. The enthalpy of liquor is 514.04 kJ/kg. Calculate the steam economy and the heating surface required.

Marks 12+8

- Q3. (a) Write on the theory of grinding in a 'Ball mill' and describe its operation with neat diagram.
- (b) What is the power required to crush 200 ton/h of quartz, if 80% of the feed passes a 2-in screen and 80% of the product a 1/8 -inch screen? The work index for quartz is 13.57.

Marks 15+5

- Q4. (a) Write on the principle, construction, operation, advantages and disadvantages of Swenson Walker crystallizer crystallizer.
- (b) A Swenson Walker crystallizer is to be used to produce 1 ton /hr of FeSO₄,7H₂O(M.W. 278) crystals by the cooling of a saturated solution (33.23%)which enters the crystallizer at T_F=125°F and the slurry leaves at T_L=75°F. After product formation, concentration of mother liquor at T_L is 22.51%. Cooling water enters counter currently the crystallizer jacket at 60°F and leaves at 70°F. Average specific heat of feed solution is 0.7 BTU/lb/°F and heat of crystallization is 28.49 BTU/lb. The overall heat transfer coefficient is 35 BTU/(hr)(sq ft)(°F). There are 3 sq ft of cooling surface per foot of crystallizer length.

(i)Estimate the cooling water requirement in litre/hr. (ii) Determine number of crystallizer units.

Marks 10+10

Q5. (a) A slurry of Calcium carbonate was filtered through a filter paper of area 0.045 m² and the solid concentration in the slurry is 24 kg/m³. Evaluate the mean specific cake resistance and the filter medium resistance at a fixed pressure of 50 kPascal. The viscosity of filtrate is 0.001 pascal. second. Use mm graph paper.

V,L	0.5	I	1.5	2	2.5	3
t,s	17.3	42.3	72	108.3	152	202.7

(b) Write short note on filter aid.

Marks 15+5

Q6. (a) A 6-bladed turbine is installed centrally in a vertical tank. The tank is 6 ft in diameter and the turbine is 2 ft in diameter. The tank is filled to a depth of 6 ft with a viscous solution of viscosity 120 centipoise and, density, 7 lb/ft³. The impellor rotates at 90 rpm.

N _{Re}	1	3	10	30	100	300	1000	3000	10000
Np	. 62	22	7.5	5	3.8	3.8	3.8	5.5	5.9

Using the given data calculate N_{Re} and determine power requirement (P) to mix this liquid?

(b) Write short note on the effects of shape factors on Np.

Marks 15+5

- Q7. Write short notes on any four of the following:
- (a)Cyclone separator, (b)Ribbon blender, (c) factors influencing blending of powder materials, (d) Basket centrifuge, (e) Trommel, (f) turbine impeller.

	Marks 4x5