

Name of Examination :B.E. CONSTRUCTION ENGINEERING FOURTH YEAR SECOND SEMESTER - 2022

Sub: ADVANCED ENVIRONMENTAL ENGINEERING(HONS.)

Marks: 42

Part-I

3+3+3=9

1. What do you mean by environment? What is pollution? Can any/all element(s) be defined as pollutants-explain in your language.
2. What are the characteristics of parameters of any waste water? Explain briefly with names and sources. Give a tentative idea of domestic waste water about its characteristic

4+8+3=15

3. How many types of treatment you know for any waste water? Draw flow sheets for primary and secondary treatments.

OR

4+5=9

4. Give details of a activated sludge plant .
5. What are the sources of air pollution? What is hazardous waste and give their basic characteristics.

9

3+6=9

**B. E. CONSTRUCTION ENGINEERING 4<sup>TH</sup> YEAR 2<sup>ND</sup> SEMESTER**  
**EXAM. 2022**

**SUBJECT: ADVANCED ENVIRONMENTAL ENGINEERING (HONS.)**

**PART - II**

Full Marks : 28

Answer any TWO questions.

	No. of Questions		Marks											
CO3	Q1.a)	What do you mean by alkalinity of water? Explain giving necessary equations.	04											
	Q1.b)	The result of chemical analysis of a sample of raw water is given below: $Ca^{++} = 60 \text{ mg/L as } CaCO_3,$ $Na^+ = 2.7 \text{ meq/L}$ Total alkalinity = 80 mg/L as $CaCO_3$ Total hardness = 120 mg/L as $CaCO_3$ $SO_4^{--} = 20 \text{ mg/L as } CaCO_3$ $Cl^- = 140 \text{ mg/L as } CaCO_3,$ $NO_3^- = 15 \text{ mg/L as } CaCO_3$ i) Prepare a bar diagram for raw water. ii) Estimate the quantity in kg /day of CaO [85% pure] and soda ash [90% pure] required to soften 3ML/day of this water.	10											
CO3	Q2.a)	Illustrate the disinfecting action of chlorine by giving the relevant equations.	4											
	Q2.b)	Results of chlorine demand test on a raw water are given below: <table border="1" style="margin-left: 20px;"> <thead> <tr> <th>Sample No.</th> <th>Chlorine dosage, mg/L</th> <th>Residual chlorine after 10 min. contact (mg/L)</th> </tr> </thead> <tbody> <tr> <td>1</td> <td>0.3</td> <td>0.18</td> </tr> <tr> <td>2</td> <td>0.5</td> <td>0.38</td> </tr> <tr> <td>3</td> <td>0.7</td> <td>0.51</td> </tr> </tbody> </table>	Sample No.	Chlorine dosage, mg/L	Residual chlorine after 10 min. contact (mg/L)	1	0.3	0.18	2	0.5	0.38	3	0.7	0.51
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**PART - II**

Full Marks : 28

Answer any TWO questions.

	No. of Questions			Marks	
		4	0.9	0.46	
		5	1.1	0.20	
		6	1.3	0.40	
		7	1.5	0.60	
		8	1.7	0.80	
	Q.3(a)	Sketch a chlorine demand curve. What is the 'break-point dosage' and what is the 'chlorine demand' at dosage of 1.3 mg/L?			6
	Q.3(b)	How many types of filters are there? Compare the salient features of SSF and RSF.			8
		Determine the quantity of alum required in order to treat 15 MLD at a treatment plant where 14 ppm of alum dose is required. Also determine the amount of CO <sub>2</sub> gas that will be liberated in this entire process.			8