

**B.E. Civil Engineering (4<sup>th</sup> Year, 2<sup>nd</sup> Semester Examination), 2022**(1<sup>st</sup> / 2<sup>nd</sup> Semester / Repeat / Supplementary / Annual / Biannual)**SUBJECT: HAZARDOUS WASTE AND ITS DISPOSAL (Elective-G)**

Full Marks: 70

Time: Two hours/Three hours/Four hours/ Six hours

(35 marks for this part)

Use a separate Answer-Script for each part

No. of Question	Part-I	Marks												
	<b><u>Answer Question-1 and any two from the rest</u></b>													
Q.1) a)	What are the key issues likely to be addressed in connection with effective hazardous waste management in our country?	(5)												
b)	What do you mean by "Partitioning of Hazardous Contaminants"? What are the different modes of partitioning of hazardous waste to various media?	(2+3)												
c)	Write short notes on i) LD <sub>50</sub> ii) Bioconcentration Factor (BCF)	(3+2)												
Q.2) a)	Determine the lower flammability limit (LFL) and upper flammability limit (UFL) of a gaseous mixture (v/v) of 0.63% Acetone, 0.22% Decane and 0.37% Hexane. Given, the LFL and UFL of individual gaseous constituents as follows:	(5)												
	<table border="1"> <thead> <tr> <th>Compound</th> <th>LFL (%)</th> <th>UFL (%)</th> </tr> </thead> <tbody> <tr> <td>Acetone</td> <td>2.6</td> <td>12.8</td> </tr> <tr> <td>Decane</td> <td>0.8</td> <td>9.2</td> </tr> <tr> <td>Hexane</td> <td>1.1</td> <td>7.5</td> </tr> </tbody> </table>	Compound	LFL (%)	UFL (%)	Acetone	2.6	12.8	Decane	0.8	9.2	Hexane	1.1	7.5	
Compound	LFL (%)	UFL (%)												
Acetone	2.6	12.8												
Decane	0.8	9.2												
Hexane	1.1	7.5												
b)	Determine the Henry's Law constant of toluene at 20°C. Vapour pressure is given to be 22 mm Hg and solubility is 512 mg/L.	(5)												
Q.3)	Determine the chronic daily intake (CDI) of a non-carcinogenic chemical in water, given that the concentration is 5.5 mg/L. Compare the CDI for an adult and child (both carcinogenic and non-carcinogenic risks involved). Given the following parameters: $ED = 30 \text{ yrs}, BW = 70 \text{ kg}, EF = 365 \frac{\text{days}}{\text{yr}}, CR = 2 \frac{\text{L}}{\text{day}}$ for adult, $CR = 1 \frac{\text{L}}{\text{day}}$ for child	(10)												
Q.4) a)	What are the points to be specifically examined during an exposure pathway assessment?	4												
b)	What do you mean by "Threshold Limit Value (TLV)"? Study in mice showed a NoAEL of 6 mg/kg for a certain contaminant. What do you think the Acceptable Daily Intake (ADI) and Reference Dose (RfD) would be?	(2+2+2)												

Ref. No. : Ex/CE/PE/B/T/421G/2022

**B.E. CIVIL ENGINEERING FOURTH YEAR SECOND SEMESTER EXAM 2022**

**SUB: HAZARDOUS WASTE AND ITS DISPOSAL**

Time: 4 hours

Full Marks: 70

Use separate answer scripts for each part

**Part-II**

**[Answer Question No.1 and any two from the rest]**

Full Marks: 35

- 1) A groundwater supply has been contaminated with ethyl benzene whose maximum level in the groundwater is 1 mg/ltr. This level is to be reduced to 35 µg/ltr by an air stripping column given the following data:

- i)  $k_1 a = 0.015 / \text{sec}$
- ii)  $\text{Liquid flow rate} = 7.5 \text{ ltr/sec}$
- iii)  $\text{temperature } (t) = 20^\circ\text{C}$
- iv)  $\text{Henry's constant} = 6.44 \times 10^{-3} \text{ atm.m}^3/\text{gm.mol}$
- v)  $\text{Column dia} = 0.6 \text{ m}$
- vi)  $\text{Air to water ratio, } \frac{Q_a}{Q_w} = 20$

Determine the liquid loading rate, stripping factor, height of the transfer unit, number of the transfer unit and height of packing in column. [2+2+2+2+3=11]

- 2) Waste stream contains 90 kg of cyanide ( $\text{CN}^-$ ) daily. Determine the stoichiometric amount of chlorine and caustic soda required to oxidize:

- i) Oxidation of Cyanide to Cyanet
- ii) Complete oxidation of cyanide to nitrogen.

Ignore the amount of caustic soda required for maintaining pH of 10. [6+6=12]

- 3) An electroplating plant generates 1600 m<sup>3</sup>/day of nickel bearing wastes having average Ni Concentration of 15000 mg/ltr as NiSO<sub>4</sub>. Assume the following characteristic of the system:

- i. Resistance through the unit = 10.5  $\Omega$ .
- ii. Current efficiency = 90%
- iii. Maximum  $\left(\frac{C.D.}{N}\right) = 5800 \text{ amp/m}^2/\text{gm-eqv/ltr}$
- iv. Membrane area = 0.95 m<sup>2</sup>

Provide a preliminary design of the system to produce 93% removal of nickel.

Determine the number of membrane, power required for the system. [12]

4. A) Describe with the help of a neat sketch the fundamentals of air stripping process involving air stripping tower.

B) Describe with the help of a neat sketch the operation of a reverse osmosis process.

[6+6=12]

5. A) Discuss on the role of ozone as chemical oxidant in hazardous waste treatment.

B) State the differences between different 'pressure membrane process' from the view point of applied pressure and size of the particles removed.

C) What is the essence of "two-film theory" in the context of air stripping process?

[5+4+3=12]