M. Sc. Chemistry Examination, 2022

(4th Semester)

ORGANIC CHEMISTRY SPECIAL

PAPER - XIII-O

Time: Two hours Full Marks: 50

(25 marks for each unit)

Use a separate answer script for each Unit.

UNIT: 0-4131

1. Predict the product(s) with proper stereochemistry in the following reactions and explain their formation with plausible mechanism (answer any *four*): 4×4

e)
$$\begin{array}{c}
 & \text{OMe} \\
 & \text{i) HCIO}_4\text{ CHCl}_3 \\
 & \text{ii) Pd-C, H}_2\text{,EtOH} \\
 & \text{iii) Et}_3\text{O}^+\text{BF}_4^-
\end{array}$$
f)
$$\begin{array}{c}
 & \text{N}_2\text{HC} \\
 & \text{MeO}
\end{array}$$

$$\begin{array}{c}
 & \text{CF}_3\text{COOH} \\
 & \text{MeO}
\end{array}$$

- Discuss the drawbacks of the use of reagent Bu₃SnH 2. a) in radical cyclization reaction. Give an example of environment friendly reagent which can be used in place of Bu₃SnH. 2+1
 - Identify the major product with probable mechanism. Discuss the role of HMPA in the reaction given below. 2+1

Predict the products with plausible mechanism. Identify the major product and justify your answer.

iii)
$$O_2N$$
 and O_2N and O_2N

- Draw the preferred mode of π -stacking interaction 6. a) between two benzene molecules. 2
 - b) How will you synthesize a coumarin derivative using an anion- π based supramolecular catalyst? 2
 - Write down a suitable template directed synthetic route of the following catenane molecule? 2

of 4-substituted-2, 6-dimethylbenzoyl chlorides in moist CH₃CN was +1.20. However, when the reaction was conducted in HClO₄ the ρ value was found as -3.90. Explain the mechanism of the above two hydrolysis reactions.

- d) Dissociation of 4-substituted phenylphosphonic acid exhibits ρ value 0.76 in H₂O at 25°C and 0.99 in 50% EtOH at 25°C. Why the ρ value increases from H₂O to 50% EtOH?
- 4. What are mechanically interlocked molecules? What are the strategies will you use to synthesize rotaxanes? Give one example of cyclodextrin-based rotaxane molecule. Draw the structure of a pH-sensitive molecular shuttle.

5. What types of noncovalent interactions present in the following molecules? 1+1+1+1

UNIT: 0-4132

3. a) Base-catalyzed hydrolysis of the following reaction exhibits 'r=0.5' and ' $\rho=+3.52$ '. Write down Yukawa-Tsuno equation and explain the hydrolysis mechanism. 2+1

b) Write down the mechanism for the product formation of the following reactions with the help of ρ value. 2×2

i)
$$OEt$$

$$EtOH$$

$$ZS^{\circ}C$$

$$OEt$$

$$OET$$

$$OET$$

$$OET$$

$$OET$$

$$OET$$

$$OMe$$

$$OMe$$

$$OOMe$$

$$OOD$$

$$O$$

The ρ value obtained from a σ -plot of the hydrolysis