

**M. SC. CHEMISTRY EXAMINATION, 2022**

( 4th Semester )

**PHYSICAL CHEMISTRY SPECIAL**

**PAPER – XVI-P**

Time : Two hours

Full Marks : 50

**(25 marks for each unit)**

**Use a separate answer script for each Unit.**

**UNIT: P-4161**

1. a) Napthalene molecule with the ten  $p_z$ -orbitals of carbon atoms as bases may be represented as the sum of irreducible representations  $A_u$ ,  $B_{1u}$ ,  $B_{2g}$ , and  $B_{3g}$  – Justify.

Using projection operator technique construct one symmetry-adapted normalized  $\pi$ MO wave function belonging to  $A_u$  symmetry. 4+5

OR

Benzene molecule with the six  $p_z$ -orbitals of carbon atoms as bases may be represented as the sum of irreducible representations  $A$ ,  $B$ ,  $E_1$  and  $E_2$  – Justify.

Using projection operator technique construct one symmetry-adapted normalized  $\pi$ MO wave function belonging to  $E_1$  symmetry. 4+5

[ 2 ]

- b) Assign the symmetry of the pure normal modes of  $\text{NH}_3$  through internal coordinate method. Show which of them are IR and Raman active. 4
2. a) Construct  $sp^2$  hybrid orbitals of  $\text{CO}_3^{2-}$  ion which belong to the  $D_{3h}$  point group. 4

Or

Find out which atomic orbitals of the atom A hybridize to form  $\sigma$  bonds with B for a molecule  $\text{AB}_5$  belonging to  $C_{4v}$  point group. 4

- b) Give reasons on the basis of symmetry why thermal and photo-chemical bond breaking of cyclobutene derivatives produce different stereo isomers. 4
- c) Show how the degenerate set of five d orbitals of a free metal atom split under an octahedral environment of ligands in a complex. 4
- [ **Note:** Character Table for required point groups will be supplied at the time of examination. ]

**UNIT: P-4162**

*Answer all the questions.*

3. Consider a free electron gas in three dimensions and hence derive an expression for the electron velocity at the Fermi surface. Comment on the factor(s) that will affect it. 6

[ 3 ]

4. What is Hall Effect? Deduce the necessary expression(s) to calculate the carrier concentration. 5
5. For alkali halide crystals like NaCl, two values of the activation energy for self-diffusion are observed – explain. 3
6. Assuming the two-sublattice model, derive an expression for the susceptibility of an antiferromagnetic material when  $T > T_N$ . 5
7. Justify and draw the qualitative energy level diagrams of a metal (M) and a p-type semiconductor (S) contact when  $\phi_M > \phi_S$  ( $\phi$  being the work function) before and after equilibrium has been established. What would be the nature of such a contact?  $3\frac{1}{2}$
8. Write a short note on (any **one**):  $2\frac{1}{2}$
- i) Piezoelectric effect
- ii) dc Josephson effect