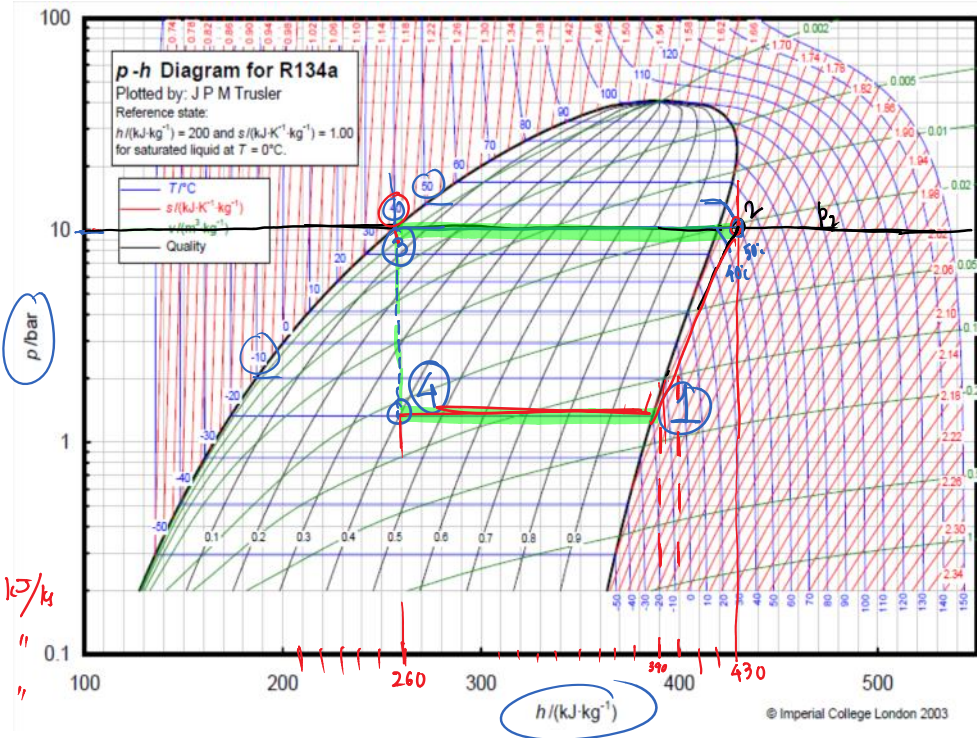
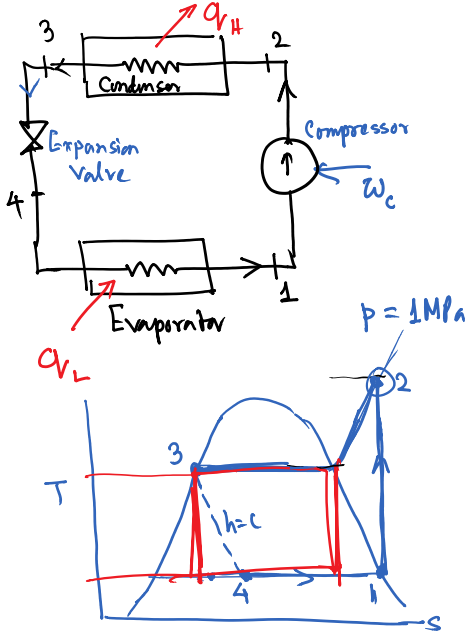


# Vapor Compression Refrigeration

A refrigerator with R-134a as the working fluid has a minimum temperature of  $-10^{\circ}\text{C}$  and a maximum pressure of  $1\text{ MPa}$ . Assume an ideal refrigeration cycle. Find the specific heat transfer from the cold space and that to the hot space, and determine the COP.



$$\begin{aligned}
 q_L &= (h_1 - h_4) = 130 \text{ kJ/kg} & h_1 &= 390 \text{ kJ/kg} \\
 w_c &= (h_2 - h_1) = 40 \text{ kJ/kg} & h_2 &= 430 \text{ kJ/kg} \\
 q_H &= (h_2 - h_3) = 170 \text{ kJ/kg} & h_4 &= 266 \text{ kJ/kg}
 \end{aligned}$$

$$\text{COP} = \frac{136}{40} = 3.25$$

$$T_L = 263 \text{ K}, \quad T_H \Big|_{\text{at } 1 \text{ MPa}} = 273 + 40 = 313 \text{ K}$$

$$(\text{COP})_{\text{Carnot}} = \frac{T_L}{T_H - T_L} = \frac{263}{313 - 263} = \boxed{5.26}$$