

Internal Combustion Engines

Lecture-4

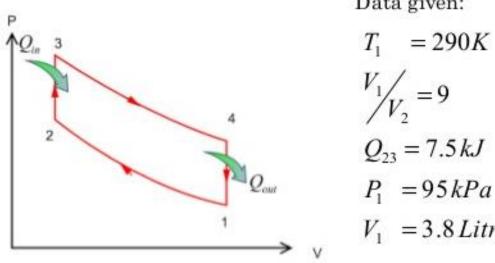
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Example Problem

An Otto cycle having a compression ratio of 9:1 uses air as the working fluid. Initially $P_1 = 95$ kPa, $T_1 = 17^{\circ}$ C, and $V_1 = 3.8$ During the heat addition process, 7.5 kJ of heat are liters. added. Determine all Ts, Ps, η_{th} , the back work ratio and the mean effective pressure.

Solution:



Data given:

$$T_{1} = 290K$$

$$V_{1}/V_{2} = 9$$

$$Q_{23} = 7.5 kJ$$

$$P_{1} = 95 kPa$$

$$V_{1} = 3.8 Litres$$

$$\frac{\Pr ccess 1 - 2(isentropic compression)}{T_{\frac{7}{2}} = \left(\frac{V_1}{V_2}\right)^{k-1}} \Rightarrow T_2 = 290(9)^{0.4} = 698.4K$$

$$\frac{P_2}{P_1} = \left(\frac{V_1}{V_2}\right)^{k-1} \Rightarrow P_2 = 95(9)^{1.4} = 2059kPa$$

$$\frac{\Pr ccess 2 - 3(Const. volume heat addition)}{P_1 t^1 law: Q_{net} - W_{net}^{-0}} = \Delta U$$

$$Q_{23} = mC_v(T_3 - T_2)$$

$$IGL: P_1v_1 = RT_1 \Rightarrow v_1 = \frac{0.2871(290)}{95} = 0.875\frac{m^3}{kg}$$

$$\frac{P_4}{P_3} = \left(\frac{V_3}{V_4}\right)^{k-1} \Rightarrow P_4 = P_3(1/9)^{1.4} = 422kPa$$

Process 4-1 (Const.volume heat rejection)

$$\begin{aligned} Q_{41} &= mC_v \left(T_4 - T_1 \right) \\ q_{41} &= C_v \left(T_4 - T_1 \right) \\ &= 0.718 \left(1288.8 - 290 \right) \\ &= 717.1 \frac{kJ}{kg} \end{aligned}$$

Then:

$$\begin{split} W_{net} &= q_{in} - q_{out} \\ &= q_{23} - q_{41} \\ &= 1009.6 \frac{kJ}{kg} \\ \eta_{th,Otto} &= \frac{W_{net}}{q_{in}} = \underbrace{0.585(58.5\%)}_{} \end{split}$$



What else?

$$MEP = \frac{W_{net}}{V_{max} - V_{min}} = \frac{W_{net}}{v_{max} - v_{min}}$$
$$= \frac{W_{net}}{v_1 - v_2} = \frac{W_{net}}{v_1 (1 - v_2 / v_1)}$$
$$= \frac{W_{net}}{v_1 (1 - \frac{1}{r})} = \frac{1009.6}{0.875(1 - 1/9)} = \underline{1298 \, kPa}$$
$$r_{bw} = \frac{W_{compr}}{W_{expans}} = \frac{\Delta u_{12}}{-\Delta u_{34}} = \frac{9 \sqrt{r} (T_2 - T_1)}{9 \sqrt{r} (T_3 - T_4)}$$
$$= \underline{0.225(22.5\%)}$$

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Example Problem 2



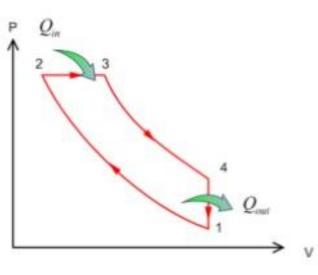
An air-standard Diesel cycle has a compression ratio of 18 and a cut-off ratio of 2.5. The state at the beginning of compression is fixed by P = 0.9 bar ant T = 300K. Calculate:

- i. the thermal efficiency of the cycle,
- ii. the maximum pressure, P_{max} , and
- iii. The mean effective pressure.

Solution:

Data given:

 $\frac{V_1}{V_2} = 18$ $\frac{V_3}{V_2} = 2.5$



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$$\Pr{ocess1-2(isentropic compression)}$$

-L = 1

$$\frac{T_2}{T_1} = \left(\frac{V_1}{V_2}\frac{1}{J}\right)^{n-1} \implies T_2 = 300(18)^{0.4} = 953.3K$$

Process 2-3(Const. pressure heat addition)

$$P_2 = P_3 \Longrightarrow \frac{V_2}{T_2} = \frac{V_3}{T_3} \Longrightarrow T_3 = T_2 \left(\frac{V_3}{V_2}\right) = 2383.3 K$$

 $\Pr ocess 3 - 4(isentropic \exp ansion)$

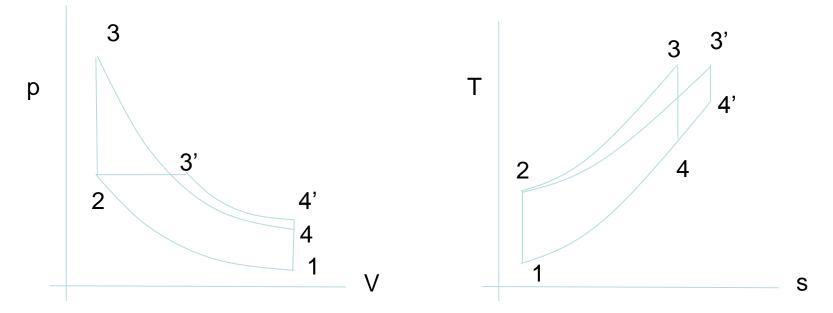
$$\frac{V_4}{V_3} = \frac{V_1}{V_2} \cdot \frac{V_2}{V_3} = 18(1/2.5) = 7.2$$
$$\frac{T_4}{T_3} = \left(\frac{V_3}{V_4}\right)^{k-1} \implies T_4 = 2383.3(1/7.2)^{0.4} = 1082 K$$

$$\begin{split} Q_{in} &= Q_{23} = mC_P \left(T_3 - T_2 \right) \Rightarrow \ q_{in} = C_P \left(T_3 - T_2 \right) = 1437.15 \frac{kJ}{kg} \\ Q_{out} &= Q_{41} = mC_P \left(T_4 - T_1 \right) \Rightarrow \ q_{out} = C_P \left(T_4 - T_1 \right) = 561.48 \frac{kJ}{kg} \\ w_{net} &= q_{in} - q_{out} = 875.67 \frac{kJ}{kg} \end{split}$$

What we need?

$$(i) \eta_{th,diesel} = \frac{w_{net}}{q_{in}} = \underbrace{0.6093(60.93\%)}_{q_{in}}$$
$$(ii) P_{max} = P_2 = P_3$$
$$\left(\frac{P_2}{P_1}\right) = \left(\frac{T_2}{T_1}\right)^{\frac{k-1}{k}} \Longrightarrow P_2 = \underbrace{5148 \, kPa(P_{max})}_{q_{max}}$$
$$(iii) MEP = \frac{w_{net}}{V_1(1-1/r)} = \frac{875.67}{0.9566(1-1/18)} = \underbrace{969.1 \, kPa}_{q_{max}}$$

Comparison of Otto and Diesel Cycles with Same Heat input and Compression Ratio



- Isobars are flat on T-s. 2-3' is flat and longer
- 4'-1 is longer and heat rejection is higher in Diesel cycle
- Otto is more efficient



Thank You