

# Heating & Cooling of Electric Machines

Day 20

# ILOs – Day20

- Derive the expressions for heating and cooling in electrical machines
- Plot the heating and cooling curves in electrical machines
- Analyze the heating and cooling conditions in electrical machines

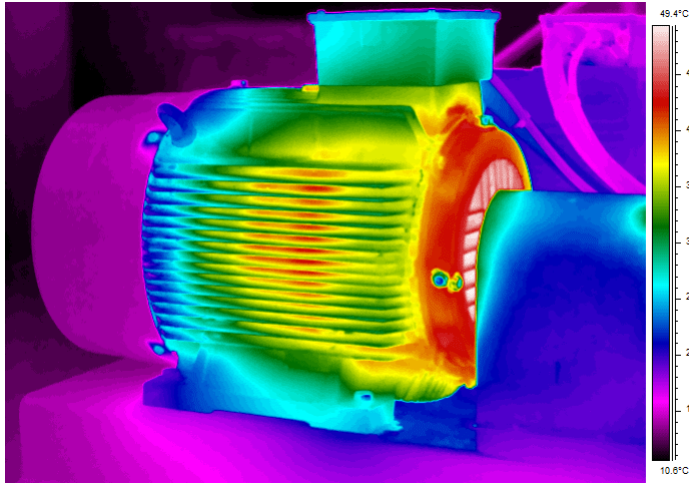
# Heating & Cooling curves

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- In an ideal **homogeneous body**, internally heated and surface cooled, has a maximum surface temperature-rise proportional directly to the power loss and inversely to that of the cooling
- Though an electric machine is far from being homogeneous, we consider it to be for the sake of simplicity
- The temperature of a machine initially rises from cold condition (=ambient temperature) as it runs with steady load
- As the temperature rises, active parts of the machine starts dissipating the heat by processes of conduction, convection, and radiation
- More the temperature rise, better will be the cooling
- The rate of temperature rise is exponential till a steady state temperature is reached

# Heating & Cooling curves

- Thermal images of heated machines



# Heating curve

# Heating curve

- Heat developed in body of the machine in small time  $dt$ :

$$Q_{dev} = Q \times dt \quad W\text{-s or } J$$

- Heat energy stored in the body during this time as the body temperature rises by  $d\theta$ :

$$Q_{st} = G \times h \times d\theta \quad J$$

- Heat dissipated during this time as the body temperature reaches  $\theta$  °C above ambient:

$$Q_{diss} = \lambda \times S \times \theta \times dt \quad W\text{-s or } J$$

- Since heat developed = heat stored + heat dissipated:

$$\Rightarrow Q \times dt = G \times h \times d\theta + \lambda \times S \times \theta \times dt$$

- $Q$  = Power loss or heat developed, W
- $G$  = Mass of the machine, kg
- $h$  = specific heat, J/kg-°C
- $S$  = cooling surface area, m<sup>2</sup>
- $\lambda$  = specific heat dissipation, W/m<sup>2</sup>-°C
- $c = 1 / \lambda$  = cooling coefficient
- $\theta$  = temperature rise at any time  $t$ , °C
- $\theta_m$  = final steady temperature rise while heating, °C
- $\theta_i$  = initial temperature rise over ambient medium, °C
- $\tau_h$  = heating time constant, s
- $\tau_c$  = cooling time constant, s
- $t$  = time, s

# Heating curve

$$Q \times dt = G \times h \times d\theta + \lambda \times S \times \theta \times dt$$

- The machine reaches steady state temperature  $\theta = \theta_m$  as  $t \rightarrow \infty$
- There is no further temperature rise

$$d\theta = 0$$

- Hence, no further energy is stored in the body of the material

$$G \times h \times d\theta = 0$$

- Amount of heat production = amount of heat dissipation

$$Q \times dt = \lambda \times S \times \theta_m \times dt$$

$$\Rightarrow \theta_m = \frac{Q}{\lambda S} = \frac{Qc}{S} = \text{The steady state temperature rise}$$

- $Q$  = Power loss or heat developed, W
- $G$  = Mass of the machine, kg
- $h$  = specific heat, J/kg- $^{\circ}\text{C}$
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- $\lambda$  = specific heat dissipation, W/ $\text{m}^2$ - $^{\circ}\text{C}$
- $c = 1 / \lambda$  = cooling coefficient
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- $t$  = time, s

# Heating curve

$$Q \times dt = G \times h \times d\theta + \lambda \times S \times \theta \times dt$$

$$\Rightarrow dt = \frac{d\theta}{\frac{Q}{Gh} - \frac{\lambda S}{Gh} \theta}$$

- Solving for the above differential equation by putting  $\theta_m = \frac{Q}{\lambda S}$  and applying suitable boundary conditions:

$$t = \frac{Gh}{\lambda S} \log_e \frac{\theta_m - \theta}{\theta_m - \theta_i}$$

$$\Rightarrow t = \tau_h \log_e \frac{\theta_m - \theta}{\theta_m - \theta_i}$$

$$\tau_h = \frac{Gh}{\lambda S} = \text{The heating time constant}$$

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- $G$  = Mass of the machine, kg
- $h$  = specific heat, J/kg- $^{\circ}\text{C}$
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\* For detailed solution please refer to appendix



# Heating curve

$$t = \tau_h \log_e \frac{\theta_m - \theta}{\theta_m - \theta_i}$$

$$\Rightarrow \frac{\theta_m - \theta}{\theta_m - \theta_i} = e^{-\frac{t}{\tau_h}}$$

$$\Rightarrow \theta = \theta_m \left( 1 - e^{-\frac{t}{\tau_h}} \right) + \theta_i e^{-\frac{t}{\tau_h}}$$

- If the machine starts from cold condition, i.e. no temperature **rise** above ambient:  $\theta_i = 0$

$$\theta = \theta_m \left( 1 - e^{-\frac{t}{\tau_h}} \right)$$

This is the *temperature rise equation with time*

- $Q$  = Power loss or heat developed, W
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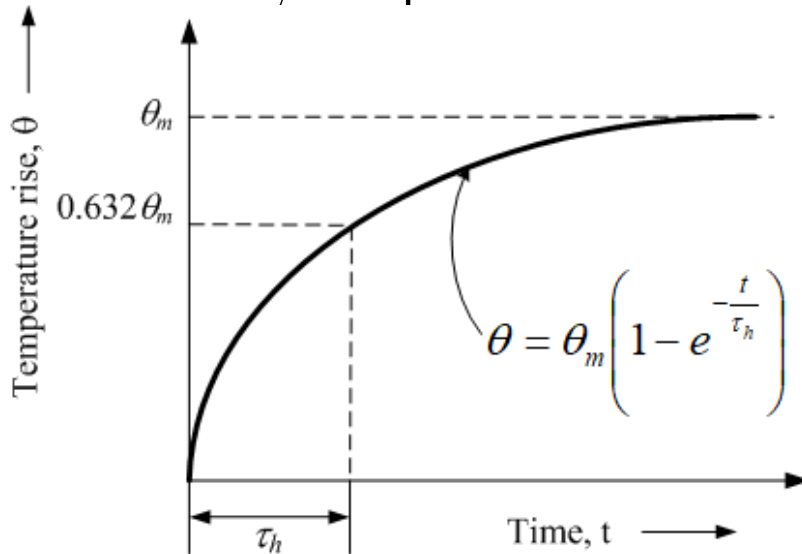
# Heating curve

$$\theta = \theta_m \left( 1 - e^{-\frac{t}{\tau_h}} \right)$$

- Putting  $t = \tau_h$  in the above equation:

$$\theta = \theta_m (1 - e^{-1}) = 0.632\theta_m$$

- Thus, the heating time constant  $\tau_h$  can be defined as the time taken by the machine to attain 63.2% of its final steady temperature rise



This is the *heating curve of a machine*

- $Q$  = Power loss or heat developed, W
- $G$  = Mass of the machine, kg
- $h$  = specific heat, J/kg- $^{\circ}$ C
- $S$  = cooling surface area,  $m^2$
- $\lambda$  = specific heat dissipation, W/ $m^2$ - $^{\circ}$ C
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- $\theta$  = temperature rise at any time  $t$ ,  $^{\circ}$ C
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# Heating curve

## Some interesting observations:

- Heating time constant:  $\tau_h = \frac{Gh}{S\lambda}$
- The time constant is inversely proportional to the specific heat dissipation  $\lambda$
- For a well ventilated machine, the value of  $\lambda$  is high and hence the time constant  $\tau_h$  is small
- Thus, a well ventilated machine will reach its steady state temperature quickly
- Conversely, the value of heating time constant is large for a poorly ventilated machine and hence it takes lot of time to reach its steady state temperature rise

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- $\lambda$  = specific heat dissipation, W/m $^2$ - $^{\circ}$ C
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- $t$  = time, s

# Heating curve

## Some interesting observations:

- Heating time constant:  $\tau_h = \frac{Gh}{S\lambda}$
- The time constant is directly proportional to volume and inversely proportional to the surface area of the machine
- Thus, overall, the heating time constant is proportional to 1<sup>st</sup> power of linear dimension
- Hence, a large sized machine will have higher value of heating time constant
- Thus, large sized machines will take more time to reach its steady state temperature

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- $\lambda$  = specific heat dissipation, W/m $^2$ - $^{\circ}$ C
- $c = 1 / \lambda$  = cooling coefficient
- $\theta$  = temperature rise at any time  $t$ ,  $^{\circ}$ C
- $\theta_m$  = final steady temperature rise while heating,  $^{\circ}$ C
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- $t$  = time, s

Conventional machines have heating time constants anywhere between half an hour to 3-4 hours

# Cooling curve

# Cooling curve

- The cooling curve of a machine can be derived from the same starting heat balance equation:

$$Q \times dt = G \times h \times d\theta + \lambda \times S \times \theta \times dt$$

- After the machine has been switched off after running for some time, it starts to cool down
- The machine reaches steady state temperature  $\theta = \theta_n$  after cooling as  $t \rightarrow \infty$
- There is no further temperature drop  
$$d\theta = 0$$
- Hence, no further energy is stored in the body of the material

$$G \times h \times d\theta = 0$$

- Amount of heat production = Amount of heat dissipation

$$Q \times dt = \lambda \times S \times \theta_n \times dt$$

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- $G$  = Mass of the machine, kg
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- $\theta_i$  = initial temperature rise over ambient medium,  $^{\circ}\text{C}$
- $\tau_h$  = heating time constant, s
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- $t$  = time, s

$$\theta_n = \frac{Q}{\lambda S}$$

# Cooling curve

- Proceeding in the same way as with heating, the expression for temperature rise with time during cooling can be derived to be:

$$\theta = \theta_n \left( 1 - e^{-\frac{t}{\tau_c}} \right) + \theta_i e^{-\frac{t}{\tau_c}}$$

Where  $\theta_n$  is the final steady state temperature rise after cooling

$$\theta_n = \frac{Q}{\lambda S}$$

$\tau_c$  is the *cooling time constant*

$$\tau_c = \frac{Gh}{\lambda S}$$

- $Q$  = Power loss or heat developed, W
- $G$  = Mass of the machine, kg
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- $t$  = time, s

*Note that all three equations are same for heating as well as cooling. However, the value of  $\lambda$  is usually different under cooling condition from that under heating condition*

# Cooling curve

$$\theta = \theta_n \left( 1 - e^{-\frac{t}{\tau_c}} \right) + \theta_i e^{-\frac{t}{\tau_c}}$$

- If the machine is shut down, no heat is produced further, and it continues to cool down till its final steady state temperature rise reaches  $0^\circ$ ; i.e.

$$\theta_n = 0$$

- Thus, temperature rise expression during cooling is:

$$\theta = \theta_i e^{-\frac{t}{\tau_c}}$$

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- $G$  = Mass of the machine, kg
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This is the *Cooling curve equation of a machine*



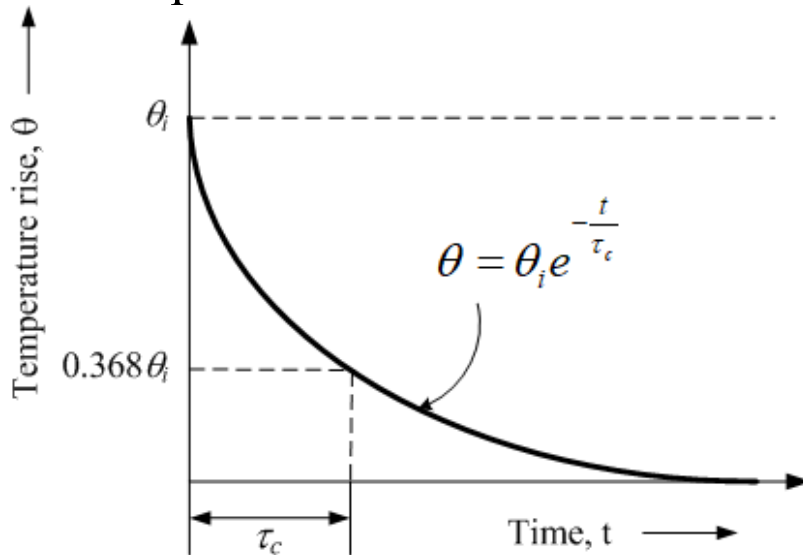
# Cooling curve

$$\theta = \theta_i e^{-\frac{t}{\tau_c}}$$

- Putting  $t = \tau_c$  in the above equation:

$$\theta = \theta_i e^{-1} = 0.368\theta_i$$

- Thus, the cooling time constant  $\tau_c$  can be defined as the time taken by the machine for its temperature rise to drop to 36.8% of its initial value



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This is the *cooling curve of a machine*

# Cooling curve

## Some interesting observations:

- As the machine is switched off, it cools down
- Thus during cooling the machine no longer rotates
- The cooling process is thus not as efficient as when it was running, and hence the value of  $\lambda$  is lower during cooling than during heating
- Thus, cooling time constant is usually larger owing to poorer ventilation condition when the machine cools
- In self-cooled motors, the cooling time constant is about 2-3 times greater than the heating time constant

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# Cooling curve

## Some interesting observations:

- Final steady state temperature rise after cooling:

$$\theta_n = \frac{Q}{\lambda S}$$

- The final steady temperature after cooling thus depends on the power loss  $Q$
- But when the machine is shut down, there is no power loss,  $Q = 0$
- Thus,  $\theta_n = 0$ , meaning that the machine temperature finally comes down to ambient temperature after cooling

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